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OPEN Resonant inelastic x-ray scattering data for Ruddlesden-Popper and DATA DESCRIPTOR reduced Ruddlesden-Popper nickelates

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Ruddlesden-Popper and reduced Ruddlesden-Popper nickelates are intriguing candidates for mimicking the properties of high-temperature superconducting cuprates. The degree of similarity between these nickelates and cuprates has been the subject of considerable debate. Resonant inelastic x-ray scattering (RIXS) has played an important role in exploring their electronic and magnetic excitations, but these efforts have been stymied by inconsistencies between different samples and the lack of publicly available data for detailed comparison. To address this issue, we present open RIXS data on La₄Ni₃O₁₀ and La₄Ni₃O₈.

Background & Summary

Generating analogs of the copper-based high-temperature superconductors based on different transition metal ions has been a major goal of materials research since the 1980s, as this opens routes to improving our understanding of unconventional superconductivity and realizing materials with improved properties^{1–3}. Since nickel lies immediately to the left of copper in the periodic table, it represents an obvious target element for realizing new superconductors. However, despite extensive work, no superconductivity has been found in nominally Ni $3d^8$ or $3d^7$ compounds⁴⁻⁶. One route to realizing the desired $3d^9$ state found in copper-based superconductors in a nickel oxide has been achieved by preparing members of the Ruddlesden-Popper phase materials with formula R_{n+1} Ni_nO_{3n+1} (where R is a rare earth element) and reducing them through removal of their apical oxygen atoms, forming R_{n+1} Ni_nO_{2n+2} materials [see Fig. 1a,b]. Indeed, this approach has been validated by the discovery of superconductivity in infinite layer $n = \infty R_{1,x} Sr_x NiO_2$, and subsequently $n = 5 Nd_6 Ni_5 O_{12}$, generating intense scientific interest^{7–10}.

Resonant x-ray spectroscopy offers several advantages for probing low valence nickelates¹¹. It can probe small sample volumes, which is useful since these materials are challenging to prepare as large single crystals. Indeed, many low valence nickelate samples are only available in thin film form. Another unique feature is the ability of RIXS to selectively study different atomic species through the resonant process and its access to magnetic and orbital excitations, which are optically forbidden¹². Efforts to understand the detailed electronic properties of

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Fig. 1 Summary of study. (**a**,**b**) Crystal structure of (**a**) La₄Ni₃O₁₀ and (**b**) La₄Ni₃O₈. La, Ni, and O atoms are depicted by green, gray, and red spheres, respectively. (**c**) The experimental geometry showing the incident x-ray angle θ_i , the scattering angle 2 Θ , and incident x-ray polarization π and σ . (**d**) An example RIXS spectrum, which encodes the electronic structure of these materials.

Ruddlesden-Popper and reduced Ruddlesden-Popper nickelates have consequently relied heavily on resonant x-ray spectroscopy^{11,13-23}.

However, much of this debate has been complicated by two major difficulties. The first is inconsistent spectra from nominally equivalent or very similar samples, an issue raised specifically in refs. ^{18,24}. The second is the lack of open data, which precludes detailed direct comparison between samples. Herein, we provide RIXS data from bulk crystals of $La_4Ni_3O_{10}$ (n=3) and $La_4Ni_3O_8$ (n=3, after reduction). In addition, little publicly accessible advice is currently available for comparing RIXS datasets, so we take this opportunity to outline considerations in comparing RIXS data taken under different experimental conditions.

Methods

The parent Ruddlesden-Popper La₄Ni₃O₁₀ was synthesized using the high pressure optical floating zone method at Argonne National Laboratory. During growth, 0.1 l/min of oxygen gas was flowed through the reaction chamber at 20 bar pressure and the feed rods were advanced at 4 mm/h over the 30 hour growth time. To improve homogeneity, the feed and seed rods were counter-rotated at 30 r.p.m. throughout this process. A similar procedure was used in ref. ²⁵. The *c*-axis surface used in the measurement was mirror polished in water using diamond lapping paper. La₄Ni₃O₈ single crystals were prepared by cleaving small crystals from the parent La₄Ni₃O₁₀ crystals and reducing them by heating in a flowing 4% H₂/Ar gas mixture at 350°C for five days. This procedure was used successfully in several prior works^{14,16,20,22,25}.

Ni $L_{3,2}$ RIXS as well as O K and Ni $L_{3,2}$ x-ray absorption spectroscopy (XAS) measurements were performed at the AERHA instrument of the SEXTANTS beamline at the SOLEIL synchrotron. The spectrometer operates by dispersing the x-ray photons as a function of their energy onto a two-dimensional detector via the detailed optical scheme described in ref. ²⁶. A sketch of the experimental geometry is displayed in Fig. 1c. All measurements were performed with the *a* and *c* sample axes in the horizontal scattering plane. At the Ni *L* edges, data were taken at $\theta_i = 15^{\circ}$ degrees (θ_i is the angle between the incoming x-ray and the sample surface), the linear dichroism is obtained by switching the incident x-ray polarization between π and σ . For the O *K* edge, a fixed π x-ray polarization was used, the dichroism is obtained by measuring at $\theta_i = 15^{\circ}$ and 90°. RIXS was measured at $2\Theta = 95^{\circ}$, with an overall full-width at half-maximum (FWHM) energy resolution of 262 ± 9 meV, chosen in order to balance throughput and resolution. As is standard in modern RIXS experiments, the two-dimensional data are binned in the isoenergetic direction to form spectra and the pixel to energy loss conversion is performed by measuring the elastic line of the spectrometer while changing the beamline energy. The data describing this calibration are provided in the Technical Validation section.

Data Records

This Data Descriptor is based on data deposited in the Zenodo general-purpose open repository²⁷. RIXS data are provided as text files with filename of the form La4Ni3O10_pi_850.0eV.dat, which specify the material, incident x-ray polarization, and incident energy. This information as well as the sample temperature are provided in the file header. XAS data are also provided as text files with a similar nomenclature, but which includes the absorption edge measured, for instance La4Ni3O8_sigma_OKedgeXAS.dat. These data are also plotted in Figs. 2–4. The XAS data displayed in Fig. 5 were taken in total fluorescence yield (TFY), but files also contain data taken in total electron yield (TEY) concomitantly. Finally, the data shown in Fig. 6a,b are stored in files elastic_850eV.dat_and pixel_energy_dispersion.dat, respectively. The repository illustrates the data content using plotting scripts based on standard python stack of numpy, matplotlib, scipy, and pandas²⁸⁻³⁰. The jupyter-repo2docker is used to define the computational environment and making the code executable on services such as mybinder.org.



Fig. 2 Ni L_3 -edge RIXS energy map for La₄Ni₃O₁₀ measured at 25 K with an incident angle of $\theta_i = 15^{\circ}$ and scattering angle of $2\Theta = 95^{\circ}$. The left and right panels show σ and π incident x-ray polarization, respectively.



Fig. 3 Ni L_3 -edge RIXS energy map for La₄Ni₃O₈ measured at 25 K with an incident angle of $\theta_i = 15^{\circ}$ and scattering angle of $2\Theta = 95^{\circ}$. The left and right panels show σ and π incident x-ray polarization, respectively.



Fig. 4 Ni L_2 -edge RIXS energy map for La₄Ni₃O₈ measured at 25 K with an incident angle of $\theta_i = 15^{\circ}$ and scattering angle of $2\Theta = 95^{\circ}$. The left and right panels show σ and π incident x-ray polarization, respectively.

Technical Validation

Figure 5 displays the Ni L_2 and O K edge XAS and x-ray linear dichroism (XLD) of La₄Ni₃O₁₀ and La₄Ni₃O₈. The larger XAS white line area seen in La₄Ni₃O₁₀ and the larger XLD in La₄Ni₃O₈ are consistent with previous results¹⁴. A similarly large XLD in La₄Ni₃O₈ is observed at the O K-edge [Fig. 5d]. These results confirm the quality of the samples used here. The energy resolution of the RIXS measurement was determined by fitting a



Fig. 5 XAS measurements of La₄Ni₃O₁₀ and La₄Ni₃O₈ taken at 300 K and $\theta_i = 15^\circ$. The Ni L_2 edge XAS and XLD are displayed in panels (**a**,**c**), respectively. Panels (**b**,**d**) display the O K edge XAS and XLD, respectively.



Fig. 6 RIXS instrument parameters. (a) Elastic line measurement; a gaussian fit yields FWHM = 262 ± 9 meV. (b) Dispersion of the elastic line through the CCD detector, the calibration follows a second order polynomial $(y=a+bx+cx^2)$ with a=912.7 eV, $b=-7.24 \times 10^{-2}$ eV/pixel, and $c=1.13 \times 10^{-5}$ eV/pixel².

Gaussian peak to a spectrum collected at an energy away from the resonance [Fig. 6a]. The data collected by the CCD area detector was converted to energy space using the calibration shown in Fig. 6b. Since the La M_4 -and Ni L_3 -edges overlap in energy, we also provide data at the Ni L_2 -edge in Fig. 4.

Usage Notes

Several technical issues make it quite difficult to measure RIXS in absolute units. We therefore present data in arbitrary units. We suggest comparing the preset spectra to other works after dividing both spectra by the integrated spectral intensity in the 0.5–15 eV energy range. It is better to exclude the elastic line from such comparisons, as the quasielastic intensity is particularly sensitive to extrinsic issues such as sample surface flatness.

To compare the data presented here with other data, care should be taken to account for possible differences in energy resolution between the different datasets. We suggest comparing data by convolving the higher-resolution spectrum with a lineshape that gives both spectra the same effective resolution.

Code availability

The data reported here were generated via synchrotron experiments and did not require any processing of datasets beyond trivial binning of the two-dimensional data into one-dimensional spectra and calibration of the energy loss.

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Author contributions

The project was conceived and supervised by G.F. & M.P.M.D. G.F., D.M., Y.S., V.B., J. F., G.S.C., A.N., N. J. & M.P.M.D. performed the x-ray measurements. J.Z. & J.F.M. synthesized the samples. G.F., D.M., Y.S., S.J., M.R.N., & M.P.M.D. performed the technical validation of the data. All authors reviewed the manuscript.

Competing interests

The authors declare no competing interests.

Additional information

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